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## 4,5-Dichlorophthalic acid-trans-cinnamamide (1/2) and 3,4,5,6-tetrachlorophthalic acid-trans-cinnamamide ( $\mathbf{1 / 2 )}$

## Hosomi, Ohba and Ito

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## 4,5-Dichlorophthalic acid-transcinnamamide (1/2) and 3,4,5,6-tetrachlorophthalic acid-trans-cinnamamide (1/2)

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In the two title adducts, $\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{Cl}_{2} \mathrm{O}_{4} \cdot 2 \mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NO}$ and $\mathrm{C}_{8} \mathrm{H}_{2} \mathrm{Cl}_{4} \mathrm{O}_{4} \cdot 2 \mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NO}$, respectively, the dicarboxylic acid is connected to two cinnamamide molecules through cyclic hydrogen bonds. The arrangement of the $\mathrm{C}=\mathrm{C}$ bonds of neighbouring cinnamamide molecules is twisted.

## Comment

The [2+2] photodimerization of trans-cinnamamides exclusively produces the $\alpha$-type photodimer (Hung et al., 1972). Photolysis of cocrystals of trans-cinnamamide with dicarboxylic acids have been investigated (Ito et al., 2000). The main photoproduct was the $\beta$-dimer of cinnamamide for phthalic acid bis(trans-cinnamamide). However, the arrangement of the $\mathrm{C}=\mathrm{C}$ bond between the neighbouring cinnamamide molecules is twisted and not suitable for [2+2] photodimerization. A pedal-like conformational change before photodimerization is expected.

The $\mathrm{C}=\mathrm{C}$ bond axes of the neighbouring cinnamamide molecules adopt a twisted arrangement in both (I) and (II), as observed in the cocrystals with phthalic acid. The distance between the centres of the $\mathrm{C} 20=\mathrm{C} 21$ and $\mathrm{C} 29=\mathrm{C} 30$ bonds is 4.453 (4) $\AA$ in (I), and that for the $\mathrm{C} 22=\mathrm{C} 23$ and $\mathrm{C} 31=\mathrm{C} 32$ bonds is 4.146 (4) $\AA$ in (II). The hydrogen-bond networks in (I) and (II) are similar to one another. Both H atoms of the $\mathrm{NH}_{2}$ groups are involved in the $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds to form a two-dimensional sheet.

Powdered crystals were spread between two Pyrex plates and irradiated with a 400 W high-pressure mercury lamp for 20 h under an argon stream at room temperature. The yields of $\beta$-dimer of cinnamamide from (I) and (II) were estimated by thin-layer chromatography and NMR spectra to be 31 and $19 \%$, respectively.

(I)

(II)

## Experimental

All the compounds were commercially available. Crystals were grown by slow evaporation from 2-propanol solution of the mixture of di-chloro- or tetrachlorophthalic acid and trans-cinnamamide (1:2).

## Compound (I)

## Crystal data

$\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{Cl}_{2} \mathrm{O}_{4} \cdot 2 \mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NO} \quad Z=2$
$M_{r}=529.38$
Triclinic, $P \overline{1}$
$a=9.6821$ (7) A
$D_{x}=1.410 \mathrm{Mg} \mathrm{m}^{-3}$
$a=9.6821$ (7) A
Mo $K \alpha$ radiation
$b=18.4661$ (12) $\AA$
Cell parameters from 25
$c=7.4203(5) \AA$ reflections
$\alpha=93.611(5)^{\circ}$
$\theta=14.3-15.0^{\circ}$
$\beta=105.676(5)^{\circ}$
$\mu=0.305 \mathrm{~mm}^{-1}$
$\begin{aligned} \beta & =10.676(5) \\ \gamma & =100.636(5)^{\circ}\end{aligned}$
$T=298$ (1) K
Plate, colourless
$V=1246.4(2) \AA^{3}$
Data collection
Rigaku AFC-7R diffractometer
$R_{\text {int }}=0.013$
$\theta-2 \theta$ scans
$\theta_{\text {max }}=27.5^{\circ}$
Absorption correction: $\psi$ scan
(North et al., 1968)
$h=-13 \rightarrow 13$
$T_{\text {min }}=0.745, T_{\text {max }}=0.955$
6174 measured reflections
5732 independent reflections
$k=-24 \rightarrow 24$
$l=-10 \rightarrow 0$
3 standard reflections every 150 reflections
4328 reflections with $I>2 \sigma(I)$
intensity decay: none

## Refinement

Refinement on $F^{2}$
All H -atom parameters refined
$R(F)=0.047$
$w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.10 P)^{2}+0.30 P\right]$
where $P=\left(F_{o}{ }^{2}+2 F_{c}{ }^{2}\right) / 3$
$w R\left(F^{2}\right)=0.198$
$(\Delta / \sigma)_{\max }=0.001$
$\Delta \rho_{\max }=0.48 \mathrm{e}_{\mathrm{A}^{-3}}{ }^{-3}$
$\Delta \rho_{\min }=-0.37 \mathrm{e}^{-3}$

5732 reflections
413 parameters
$0.70 \times 0.40 \times 0.15 \mathrm{~mm}$

Table 1
Selected geometric parameters $(\AA)$ for (I).

| C11-C15 | $1.720(3)$ | O8-C28 | $1.251(3)$ |
| :--- | :--- | :--- | :--- |
| C12-C14 | $1.731(3)$ | N9-C19 | $1.331(4)$ |
| O3-C11 | $1.204(3)$ | N10-C28 | $1.322(3)$ |
| O4-C11 | $1.317(3)$ | C19-C20 | $1.482(4)$ |
| O5-C18 | $1.307(3)$ | C20-C21 | $1.298(4)$ |
| O6-C18 | $1.206(3)$ | C28-C29 | $1.481(4)$ |
| O7-C19 | $1.252(3)$ | C29-C30 | $1.315(3)$ |

Table 2
Hydrogen-bonding geometry ( $\AA,^{\circ}$ ) for (I).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| O4-H4 $\cdots \mathrm{O} 7$ | $0.83(4)$ | $1.78(4)$ | $2.553(3)$ | $155(5)$ |
| O5-H5 | O8 | $0.84(4)$ | $1.69(4)$ | $2.491(3)$ |
| N9-H9A $\cdots$ O3 | $1.06(3)$ | $1.86(3)$ | $2.905(4)$ | $158(4)$ |
| N9-H9B (5) | 168 | $0.87(4)$ | $2.35(5)$ | $3.081(3)$ |
| N10-H10A $\cdots$ O6 | $1.01(4)$ | $1.94(4)$ | $2.923(3)$ | $142(5)$ |
| N10-H10B $\cdots$ O7 $^{\mathrm{ii}}$ | $0.78(4)$ | $2.27(4)$ | $3.039(3)$ | $175(4)$ |

Symmetry codes: (i) $-x,-y,-z$; (ii) $1-x,-y, 1-z$.

## Compound (II)

## Crystal data

| $\mathrm{C}_{8} \mathrm{H}_{2} \mathrm{Cl}_{4} \mathrm{O}_{4} \cdot 2 \mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NO}$ | $Z=2$ |
| :--- | :--- |
| $M_{r}=598.27$ | $D_{x}=1.495 \mathrm{Mg} \mathrm{m}^{-3}$ |
| Triclinic, $P \overline{1}$ | Mo $K \alpha$ radiation |
| $a=9.937(2) \AA$ | Cell parameters from 25 |
| $b=15.895(4) \AA$ | reflections |
| $c=9.015(2) \AA$ | $\theta=14.4-15.0^{\circ}$ |
| $\alpha=93.48(2)^{\circ}$ | $\mu=0.490 \mathrm{~mm}^{-1}$ |
| $\beta=110.57(1)^{\circ}$ | $T=298(1) \mathrm{K}$ |
| $\gamma=91.58(2)^{\circ}$ | Plate, colourless |
| $V=1328.8(5) \AA^{\circ}$ | $0.60 \times 0.60 \times 0.15 \mathrm{~mm}$ |

## Data collection

Rigaku AFC-7R diffractometer $\theta-2 \theta$ scans
Absorption correction: $\psi$ scan
(North et al., 1968)
$T_{\text {min }}=0.772, T_{\text {max }}=0.929$
6442 measured reflections
6092 independent reflections
4659 reflections with $I>2 \sigma(I)$

$$
\begin{aligned}
& R_{\text {int }}=0.029 \\
& \theta_{\max }=27.5^{\circ} \\
& h=-13 \rightarrow 0 \\
& k=-21 \rightarrow 21 \\
& l=-12 \rightarrow 12 \\
& 3 \text { standard reflections } \\
& \quad \text { every } 150 \text { reflections } \\
& \quad \text { intensity decay: none }
\end{aligned}
$$

## Refinement

Refinement on $F^{2}$
$R(F)=0.038$
$w R\left(F^{2}\right)=0.113$
$S=1.06$
6092 reflections
423 parameters
All H -atom parameters refined

Table 3
Selected geometric parameters ( $\AA$ ) for (II).

| $\mathrm{Cl} 1-\mathrm{C} 18$ | $1.722(2)$ | $\mathrm{O} 9-\mathrm{C} 21$ | $1.249(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{Cl} 2-\mathrm{C} 17$ | $1.720(3)$ | $\mathrm{O} 10-\mathrm{C} 30$ | $1.265(3)$ |
| $\mathrm{Cl} 3-\mathrm{C} 16$ | $1.718(2)$ | $\mathrm{N} 11-\mathrm{C} 21$ | $1.313(3)$ |
| $\mathrm{Cl} 4-\mathrm{C} 15$ | $1.725(2)$ | $\mathrm{N} 12-\mathrm{C} 30$ | $1.306(3)$ |
| $\mathrm{O} 5-\mathrm{C} 13$ | $1.208(2)$ | $\mathrm{C} 21-\mathrm{C} 22$ | $1.472(4)$ |
| $\mathrm{O} 6-\mathrm{C} 13$ | $1.312(2)$ | $\mathrm{C} 22-\mathrm{C} 23$ | $1.311(3)$ |
| $\mathrm{O} 7-\mathrm{C} 20$ | $1.294(2)$ | $\mathrm{C} 30-\mathrm{C} 31$ | $1.467(3)$ |
| $\mathrm{O} 8-\mathrm{C} 20$ | $1.210(2)$ | $\mathrm{C} 31-\mathrm{C} 32$ | $1.326(3)$ |

Table 4
Hydrogen-bonding geometry $\left({ }^{\circ},{ }^{\circ}\right)$ for (II).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| O6-H6 $\cdots$ O9 | $0.93(4)$ | $1.68(4)$ | $2.575(3)$ | $163(3)$ |
| O7-H7 O10 | $1.03(4)$ | $1.43(4)$ | $2.444(2)$ | $167(4)$ |
| N11-H11A $\cdots$ O5 | $0.81(3)$ | $2.15(3)$ | $2.949(3)$ | $169(3)$ |
| N11-H11B $\cdots$ O7 $^{\text {i }}$ | $0.89(4)$ | $2.23(4)$ | $3.120(3)$ | $177(3)$ |
| N12-H12A $\cdots$ O8 $^{\text {(ii }}$ | $0.88(3)$ | $1.97(3)$ | $2.835(3)$ | $166(2)$ |
| N12-H12B $\cdots$ O $^{2}$ | $0.88(3)$ | $2.14(3)$ | $3.001(2)$ | $167(3)$ |

Symmetry codes: (i) $1-x,-y, 1-z$; (ii) $-x,-y,-z$.
The refined bond distances involving the H atoms are 0.78 (4)1.06 (5) and 0.81 (3)-1.03 (4) $\AA$ for (I) and (II), respectively.

For both compounds, data collection and cell refinement: $\mathrm{MSC} /$ AFC Diffractometer Control Software (Molecular Structure Corporation, 1993); data reduction: TEXSAN (Molecular Structure Corporation, 1999); program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); software used to prepare material for publication: TEXSAN.

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